

# Van der Waals epitaxial MOCVD-growth of $(\text{Bi}_x\text{Sb}_{1-x})_2\text{Te}_3$ ( $0 < x < 1$ ) films

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## Abstract

Epitaxial  $(\text{Bi}_x\text{Sb}_{1-x})_2\text{Te}_3$  films with ( $0 < x < 1$ ) were grown by the metal-organic chemical vapour deposition (MOCVD) process at 400 °C using the tailor-made precursors  $\text{Et}_2\text{Te}_2$ ,  $i\text{-Pr}_3\text{Sb}$  and  $\text{Et}_3\text{Bi}$ . The films grown on  $\text{Al}_2\text{O}_3(0001)$  substrates show a very smooth surface morphology as shown by a scanning electron microscope (SEM), atomic force microscopy (AFM) and transmission electron microscopy (TEM), while those grown on Si(100) are rather polycrystalline. The chemical composition of the crystalline films (x-ray powder diffraction (XRD)) was investigated by energy-dispersive x-ray (EDX) and x-ray photoelectron spectroscopy (XPS), and the in-plane transport properties were measured, and a strong dependency from the bismuth content was found, which allows the tuning of the carrier concentration and mobility in a wide range.

Keywords: Van der Waals epitaxy, MOCVD, topological insulators, thermoelectric materials,  $(\text{Bi}_x\text{Sb}_{1-x})_2\text{Te}_3$

(Some figures may appear in colour only in the online journal)

## Introduction

Thermoelectric materials have attracted considerable attention in recent years since the thermoelectric effect allows the direct conversion of thermal energy into electrical energy and vice versa [1–5]. Systematic research in this field began in the 1950s and resulted in the development of radioisotope thermoelectric generators (RTG), which provide the energy supply of satellites and space probes, for example [6]. The energy conversion efficiency of thermoelectric materials is given by a dimensionless figure of merit  $ZT = \alpha^2 \sigma T / (\kappa_e + \kappa_L)$ , where  $\alpha$  is the Seebeck coefficient,  $\sigma$  the electrical conductivity,  $T$  the absolute temperature and  $\kappa_e$  and  $\kappa_L$  the thermal conductivities of the electron and lattice components, respectively. Consequently, a good thermoelectric material must exhibit a high Seebeck coefficient and

electrical conductivity, whereas its thermal conductivity should be as low as possible.

Presently, the best thermoelectric materials for near-room temperature applications are binary group VI-tellurides,  $\text{Sb}_2\text{Te}_3$  and  $\text{Bi}_2\text{Te}_3$ , and their ternary solid solution  $(\text{Bi}_x\text{Sb}_{1-x})_2\text{Te}_3$  ( $0 < x < 1$ ), respectively, which are narrow bandgap semiconductors with layered tetradymite-type structures [7, 8]. The structure is built by a close packing of tellurium atoms perpendicular to the  $c$ -axis in  $R\bar{3}m:HH$  with the stacking sequence  $chh$  in Jagodzinski symbols, occupying octahedral sites except in every third layer of the  $\text{Te}_6$  octahedra. The stacking sequence is  $(A\gamma B\Box A\gamma Ba C\Box Ba C\beta A\Box C\beta)$  and thus, double layers of edge-sharing  $\text{Bi}(\text{Sb})\text{Te}_6$  octahedra consisting of five atom layers (quintuple layers) with the sequence  $\text{Te1-Bi}(\text{Sb})\text{-Te2-Bi}(\text{Sb})\text{-Te1}$  are formed. Each double layer of octahedra (quintuple layer) has the composition  $(\text{Bi}(\text{Sb}))_2\text{Te}_3$ ,

which is a condition for layer structures. There are three formula units within the unit cell and so three stacks of quintuple layers are connected via weak van der Waals bonding, whereas the bonding inside the layer is dominated by mixed covalent and ionic types.

The interest in  $\text{Bi}_2\text{Te}_3$  thin films and related materials such as  $\text{Bi}_2\text{Se}_3$  and  $\text{Sb}_2\text{Te}_3$  recently received a second major boost since they are archetypes of materials referred to as topological insulators [9–11], which are characterized by the topologically protected conducting surface-states forming a single spin-polarized Dirac cone on the bulk insulating interior due to strong spin–orbit coupling [12–16]. Very recently, it was demonstrated that the topological surface states can be modulated and insulating bulk states can be achieved in the solid solution  $(\text{Bi}_x\text{Sb}_{1-x})_2\text{Te}_3$  [17].

$ZT$  values of thermoelectric materials can be enhanced by nanostructuring approaches [18–21], since the thermal conductivity in low-dimensional nanostructures is decreased due to efficient phonon scattering at boundaries and interfaces, while the Seebeck coefficient is simultaneously increased because of quantum confinement effects and the modification of the electronic band structure [22, 23]. In addition, the synthesis of solid solutions in the quasi binary system  $\text{Bi}_2\text{Te}_3$ – $\text{Sb}_2\text{Te}_3$  has been established for  $ZT$  value improvement.  $(\text{Bi}_x\text{Sb}_{1-x})_2\text{Te}_3$  compounds, for instance, show enhanced  $ZT$  values compared to the corresponding binary materials ( $\text{Bi}_2\text{Te}_3$ ,  $\text{Sb}_2\text{Te}_3$ ) due to additional phonon scattering introduced by lattice disordering (site disordering), which strongly reduces the thermal conductivity. Moreover, the formation of antisite defects, i.e. the occupation of tellurium sites by pnictogen atoms, is suppressed with an increasing bismuth content due to the enhanced electronegativity difference, which leads to a higher defect formation energy, a decreased carrier concentration, and an increased Seebeck coefficient. The transport properties of  $(\text{Bi}_x\text{Sb}_{1-x})_2\text{Te}_3$  single crystals were investigated using theoretical and experimental methods and a maximum room temperature figure of merit  $Z = 3.2 \times 10^{-3} \text{ K}^{-1}$  was determined for the solid solution [24, 25], while  $p$ -type  $(\text{Bi}_x\text{Sb}_{1-x})_2\text{Te}_3$  nanocomposites showed the maximum room temperature  $Z$  of  $3.52 \times 10^{-3} \text{ K}^{-1}$  for  $(\text{Bi}_{0.2}\text{Sb}_{0.8})_2\text{Te}_3$  [26]. Poudel *et al* reported on record-high  $ZT$  values of 1.4 at 373 K for bulk  $(\text{Bi}_x\text{Sb}_{1-x})_2\text{Te}_3$  with embedded nanostructures [1] while Xie *et al* reported on high performance  $p$ -type  $(\text{Bi}_{0.26}\text{Sb}_{0.74})_2\text{Te}_3$  material with a maximum  $ZT$  value of 1.56 at 300 K, which is a roughly 50% improvement compared to commercial  $\text{Bi}_2\text{Te}_3$  materials [27].

While the synthesis of bulk as well as nanostructured  $p$ -type  $(\text{Bi}_x\text{Sb}_{1-x})_2\text{Te}_3$  is well established [1, 28–33], reports on the growth of high quality epitaxial thin films of these materials are rather scarce [34]. Ternary  $(\text{Bi}_x\text{Sb}_{1-x})_2\text{Te}_3$  thin films with a very limited range of  $x$  values were fabricated by molecular beam epitaxy [35], physical vapor deposition [36], mechanical alloying [37, 38], dc magnetron sputtering [39] and potentiostatic electrodeposition [40].  $(\text{Bi}_{0.25}\text{Sb}_{0.75})_2\text{Te}_3$  is the best commercially available  $p$ -type thermoelectric material at room temperature and much effort has been made to improve the performance of  $(\text{Bi}_x\text{Sb}_{1-x})_2\text{Te}_3$  films [41]. However, the growth of such ternary films with specific

preferential crystal growth on the substrate is still challenging. In addition, these films often show low power factors, since the carrier concentrations in many films often range from  $10^{20}$ – $10^{21} \text{ cm}^{-3}$ , which exceed the optimal value of carrier concentration of roughly  $10^{19} \text{ cm}^{-3}$  [41] and limit the Seebeck coefficient. Moreover, ternary  $(\text{Bi}_x\text{Sb}_{1-x})_2\text{Te}_3$  material films with a broad variation of  $x$  were only scarcely realized in electrochemically synthesized  $(\text{Bi}_x\text{Sb}_{1-x})_2\text{Te}_3$  thin films [42] and by magnetron co-sputtering [43].

The metal-organic chemical vapour deposition (MOCVD) process, which has previously been established for the deposition of high quality films of binary materials including  $\text{Bi}_2\text{Te}_3/\text{Sb}_2\text{Te}_3$  superlattices by the so-called Van der Waals epitaxy [44–46] was used much less for ternary film deposition [47, 48].  $(\text{Bi}_x\text{Sb}_{1-x})_2\text{Te}_3$  films were successfully grown using  $i$ - $\text{Pr}_2\text{Te}$ ,  $\text{Me}_3\text{Bi}$  and  $\text{Et}_3\text{Sb}$  as metal organic precursors [49, 50]. Very recently, high quality  $(\text{Bi}_{0.53}\text{Sb}_{0.47})_2\text{Te}_3$  thin films with atomically sharp interfaces and without any interfacial layers were grown by van der Waals epitaxy on a GaSe buffer layer [51]. We have recently demonstrated the MOCVD growth of epitaxial  $\text{Sb}_2\text{Te}_3$  films and investigated their valence band structure by angle-resolved photoemission spectroscopy [52]. Here, we report on the deposition of  $(\text{Bi}_x\text{Sb}_{1-x})_2\text{Te}_3$  thin films over the entire range from  $\text{Bi}_2\text{Te}_3$  to  $\text{Sb}_2\text{Te}_3$  ( $x = 0 - 1$ ) at temperatures below  $400^\circ\text{C}$  using  $i$ - $\text{Pr}_3\text{Sb}$ ,  $\text{Et}_3\text{Bi}$  and  $\text{Et}_2\text{Te}_2$  as alternate metal organic precursors for low-temperature MOCVD film deposition. Low growth temperatures are advantageous in order to suppress interdiffusion in these films, which becomes problematic when using the standard Bi precursor  $\text{Me}_3\text{Bi}$  due to its rather high thermal stability. The composition and morphology of the resulting epitaxial thin films was studied in detail and their electric conductivity was determined.

## Experimental details

### MOCVD deposition

MOCVD studies were performed in a cold-wall high-vacuum MOCVD reactor.  $(\text{Bi}_x\text{Sb}_{1-x})_2\text{Te}_3$  films were deposited on either Si(100) or  $\text{Al}_2\text{O}_3(0001)$  substrates within 15 min at a working pressure of 10 mbar. Si(100) substrates were degreased with acetone, treated with hydrofluoric acid and heated to  $500^\circ\text{C}$  at  $10^{-3}$  mbar for 1 h in the reactor.  $\text{Al}_2\text{O}_3(0001)$  substrates were degreased with acetone and cleaned with a 3:1 mixture of sulfuric acid and phosphoric acid.  $\text{Et}_2\text{Te}_2$ , [53]  $i$ - $\text{Pr}_3\text{Sb}$  [54] and  $\text{Et}_3\text{Bi}$  [55], which were synthesized by methods slightly modified from those described in the literature, were loaded into glass bubblers attached to the MOCVD reactor under inert conditions (Ar). Argon was used as the carrier gas, and the flow rate of the precursors (40 sccm  $\text{Et}_2\text{Te}_2$ , 5 sccm  $i$ - $\text{Pr}_3\text{Sb}$  and 5 sccm  $\text{Et}_3\text{Bi}$ ) was controlled by use of a mass flow controller (MKS Instruments). The flow of the precursors was adjusted by the bubbler temperature ( $i$ - $\text{Pr}_3\text{Sb}$ :  $-25$  to  $-5^\circ\text{C}$ ;  $\text{Et}_3\text{Bi}$ :  $-50$  to  $-30^\circ\text{C}$ ;  $\text{Et}_2\text{Te}_2$ :  $+20^\circ\text{C}$ ). After the film deposition was

finished, the system was cooled to ambient temperature within 30 min under vacuum.

#### DSC analysis

A DSC 200 Phox (Netzsch Gerätebau) was used for differential scanning calorimetry (DSC) analyses.

#### X-ray analysis

XRD patterns were obtained using a Bruker D8 Advance powder diffractometer with Cu  $K_{\alpha}$  radiation ( $\lambda$ : 1.5418 Å).

#### X-ray photoelectron spectroscopy

XPS studies were performed using a Versaprobe II<sup>TM</sup> (Ulvac-Phi) with monochromatic Al  $K_{\alpha}$  light at 1486.6 eV photon energy. For depth-profiling this machine is equipped with an Ar-sputter source. The emission angle between analyzer and sample is 45°.

#### Surface analysis

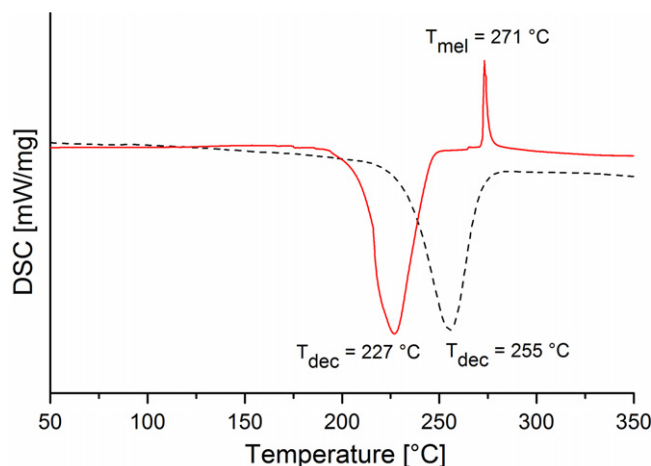
The surface morphology of the  $Sb_2Te_3$  film was investigated by atomic force microscopy (AFM) using an AFM Veeco di Innova and by scanning electron microscope (SEM) using a Jeol JSM 6510 equipped with an energy dispersive x-ray spectroscopy (EDX) device (Bruker Quantax 400). A cross section sample of the  $Sb_2Te_3$  film was prepared by use of a Jeol Cross-Section Polisher (IB-09010CP).

#### Transmission electron microscopy

TEM studies were conducted on a FEI Philips transmission electron microscope CM30 T/LaB6 operated at 300 kV, equipped with a Gatan CCD for image recording and with a Thermo NSS system for EDX analysis using a Si(Li) nanotracer detector. Cross section samples of the coated substrates prepared by focused ion beam (FIB) thinning were prepared using a FEI Helios NanoLab<sup>TM</sup> 600.

## Results and discussion

The standard tellurium precursor for MOCVD applications is dimethyltellane  $Me_2Te$ .  $Me_2Te$  is liquid at room temperature (b.p. 82 °C) and has a sufficiently high vapor pressure (40.6 Torr, 20 °C) [56], but is thermally very stable and requires a high thermolysis temperature of 500 °C in  $H_2$  atmosphere to be fully decomposed [57]. In addition, the decomposition occurs under formation of methyl radicals, a potential source for carbon contamination. The decomposition temperatures of dialkyltellanes containing longer C-chains with additional branching ( $Et_2Te$ ,  $i-Pr_2Te$ ,  $t-Bu_2Te$ ) typically decrease due to an increased stability of the resulting radicals [57, 58]. Moreover,  $\beta$ -hydride elimination reactions on the surface of the substrate have to be taken into account [59–61]. In order to achieve the lowest possible substrate temperature to prevent the loss of tellurium from the film, we used



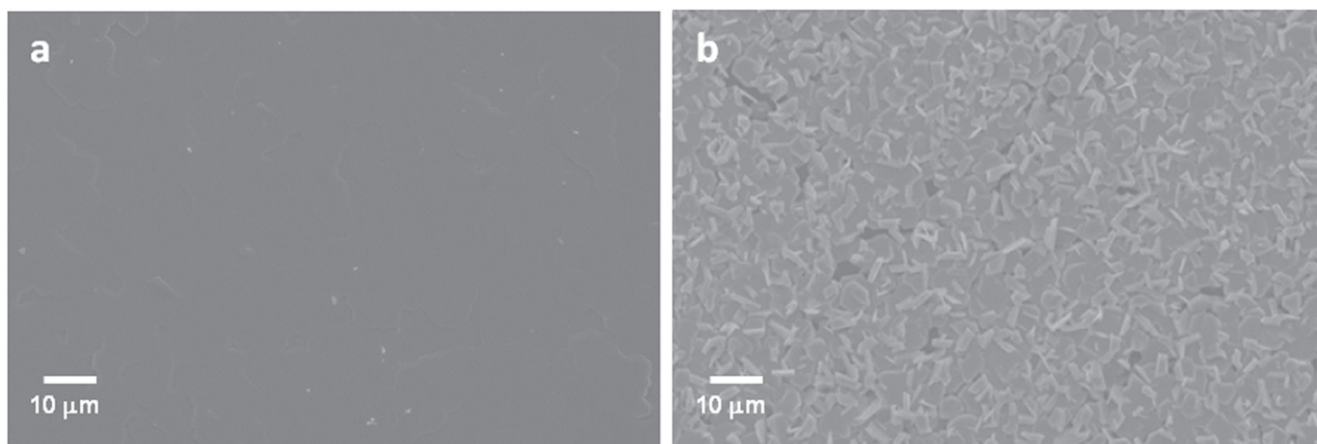
**Figure 1.** DSC curves of  $i-Pr_3Sb$  (red curve) [52] and  $Et_3Bi$  (black curve).

diethyltellane  $Et_2Te_2$  as the Te source and reported on its thermal properties in our previous report [52]. It is suggested that  $Et_2Te_2$  decomposes in a two-step mechanism, as was also reported by Kisker *et al* for  $Me_2Te_2$  [62]. First  $Et_2Te_2$  disproportionates into elemental tellurium and diethyl tellane, which then thermally degrades at a higher temperature, leading to a high tellurium concentration at the growth front.  $i-Pr_3Sb$  and  $Et_3Bi$  were used as Sb and Bi precursors due to their comparable decomposition temperatures of roughly 250 °C and 230 °C (figure 1), respectively.

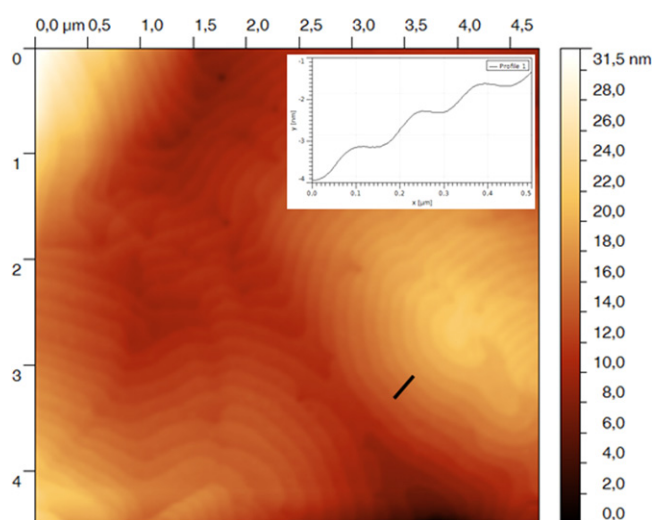
$(Bi_xSb_{1-x})_2Te_3$  films were deposited at 400 °C on  $Al_2O_3(0001)$  substrates. They are typically highly reflective and the SEM images show the formation of dense, flat films over the entire range of the  $Al_2O_3$  substrate (figure 2(a)). In contrast, the films deposited on  $Si(100)$  at 400 °C are dull and grey and extremely rough according to SEM studies. They consist of hexagonal platelets lying either flat to the substrate or standing perpendicular on the surface (figure 2(b)).

Binary antimony and bismuth telluride  $M_2Te_3$  ( $M = Sb, Bi$ ) as well as the ternary tellurides are described in the space group  $R-3m:H$  (No. 166) and form layered, rhombohedral crystal structures built by anisotropic layers, in which five atomic planes are covalently bonded to form a  $Te(1)-M-Te(2)-M-Te(1)$  quintuple layer. The conventional unit cell includes three quintuple layers, which are interconnected by weak van der Waals forces. An AFM micrograph of a  $(Bi_{0.5}Sb_{0.5})_2Te_3$  film grown on  $Al_2O_3(0001)$  shows regular terraces 1 nm in height (figure 3), which corresponds to one quintuple layer of the layered tetradymite-type structure of  $M_2Te_3$ , that is roughly 1 nm thick. These findings confirm a layered growth, as was previously observed in molecular beam epitaxy (MBE) grown  $M_2Te_3$  films [63].

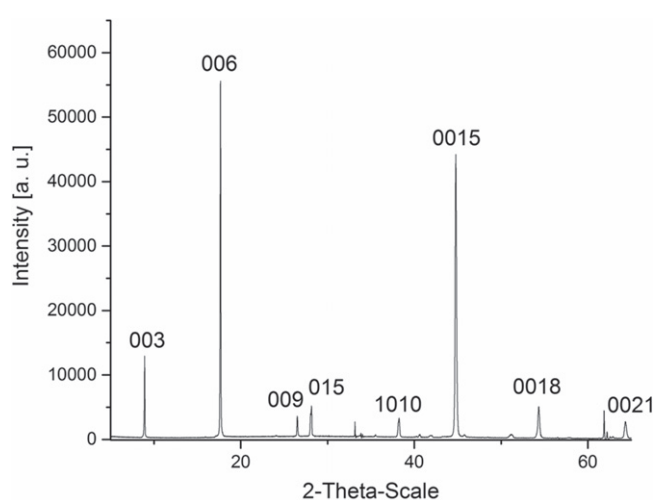
The chemical composition of the films was measured by EDX analysis. The measurements clearly showed that the composition for each film is close to the ideal 40:60 (Sb/Bi:Te) molar ratio. Figure 4 displays the XRD pattern, measured in  $\theta-2\theta$  (Bragg–Brentano) set-up for the MOCVD grown films on  $Al_2O_3(0001)$ .



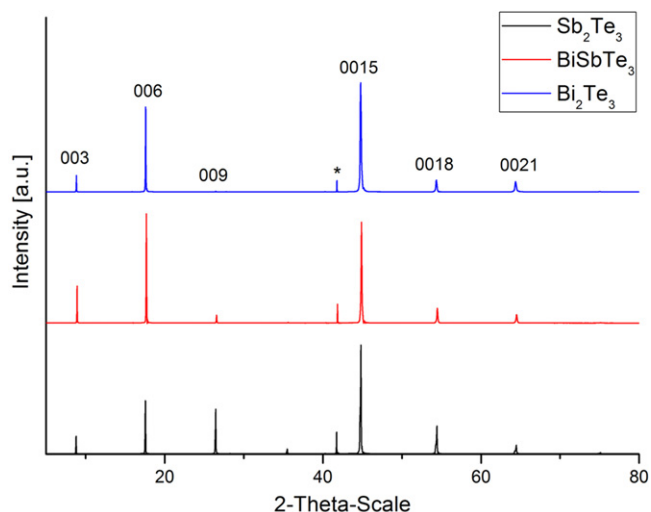
**Figure 2.** SEM images of  $(\text{Bi}_{0.5}\text{Sb}_{0.5})_2\text{Te}_3$  films grown on  $\text{Al}_2\text{O}_3(0001)$  (a) and  $\text{Si}(100)$  (b) at  $400^\circ\text{C}$ .



**Figure 3.** AFM photograph of a  $(\text{Bi}_{0.5}\text{Sb}_{0.5})_2\text{Te}_3$  film grown on  $\text{Al}_2\text{O}_3(0001)$  at  $400^\circ\text{C}$ .



**Figure 5.** X-ray diffractogram of a  $(\text{Bi}_{0.5}\text{Sb}_{0.5})_2\text{Te}_3$  film grown on  $\text{Si}(100)$  at  $400^\circ\text{C}$ .

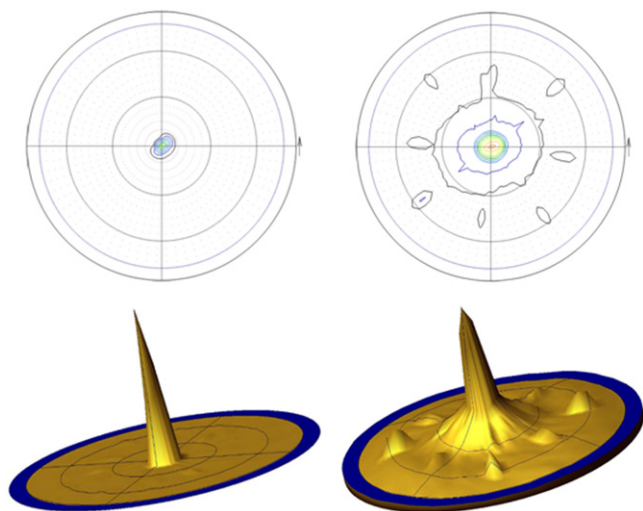


**Figure 4.** X-ray diffractograms of  $\text{Sb}_2\text{Te}_3$  (black),  $\text{Bi}_2\text{Te}_3$  (blue) and  $(\text{Bi}_{0.5}\text{Sb}_{0.5})_2\text{Te}_3$  films (red) grown on  $\text{Al}_2\text{O}_3(0001)$  at  $400^\circ\text{C}$ ; the marked reflex (\*) corresponds to the substrate  $\text{Al}_2\text{O}_3(0001)$ .

All observed Bragg peaks can be indexed on the basis of rhombohedral  $\text{Sb}_2\text{Te}_3$  (JCPDS card 15-0874) or  $\text{Bi}_2\text{Te}_3$  (JCPDS card 15-0863), respectively. The small full width at half maximum (FWHM) of the peaks indicates a high degree of crystallinity. Only diffraction peaks related to the 00/ crystal planes are visible, whereas compared to the standard card other peaks are totally suppressed, indicating the growth direction to be perfectly aligned along the (00/ crystal orientation with the *c*-axis of the tetradymite lattice perpendicular to the substrate surface. No peaks for other phases or impurities such as  $\text{Bi}_2\text{O}_3$ ,  $\text{Sb}_2\text{O}_3$ ,  $\text{TeO}_2$  or elemental Te were detected.

In contrast, the XRD pattern for  $(\text{Bi}_x\text{Sb}_{1-x})_2\text{Te}_3$  films grown on  $\text{Si}(100)$  substrates unexceptionally show strong texture effects, as is shown for a  $(\text{Bi}_{0.5}\text{Sb}_{0.5})_2\text{Te}_3$  film (figure 5). The diffractogram shows enhanced peak intensities for the 00/ reflexes, but other peaks are clearly visible and confirm the two preferred orientations of the platelets in the film as observed by SEM.

Figure 6 shows the 006 pole figures of  $(\text{Bi}_{0.5}\text{Sb}_{0.5})_2\text{Te}_3$  films grown on  $\text{Al}_2\text{O}_3(0001)$  (left) and  $\text{Si}(100)$  (right) at



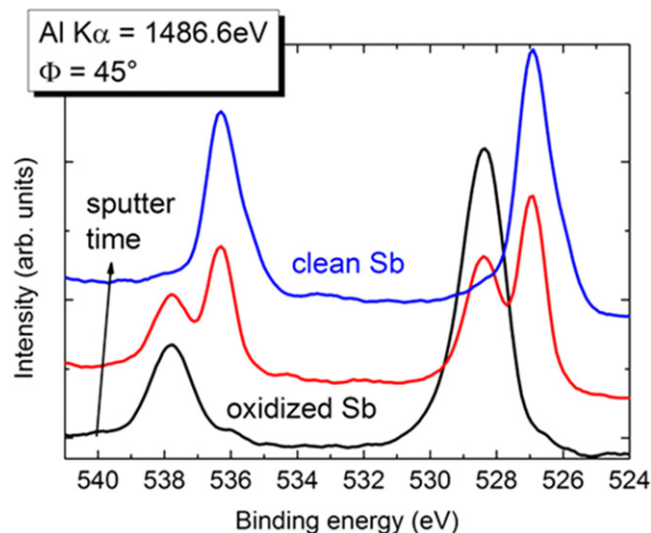
**Figure 6.** 006 pole figures for  $(\text{Bi}_{0.5}\text{Sb}_{0.5})_2\text{Te}_3$  films grown on  $\text{Al}_2\text{O}_3(0001)$  (left) and  $\text{Si}(100)$  (right).

400 °C. The epitaxial  $(\text{Bi}_{0.5}\text{Sb}_{0.5})_2\text{Te}_3$  film grown on  $\text{Al}_2\text{O}_3$  only shows a single peak, while the 0006 pole figure of the film grown on Si shows a pattern due to the hexagonal discs standing on the substrate surface. The epitaxial relationship of the film and the  $\text{Al}_2\text{O}_3$  substrate is  $(\text{Bi}_{0.5}\text{Sb}_{0.5})_2\text{Te}_3(001) \parallel \text{Al}_2\text{O}_3(0001)$ .

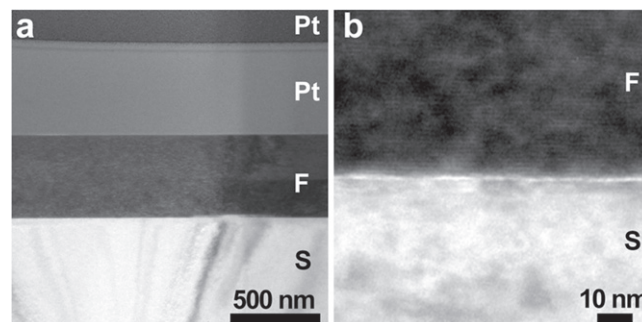
X-Ray photoelectron spectroscopy (XPS) is a surface sensitive method for the determination of the chemical environment of an element. An XPS measurement for a  $(\text{Bi}_{0.5}\text{Sb}_{0.5})_2\text{Te}_3$  film grown at 400 °C on  $\text{Al}_2\text{O}_3(0001)$  shows the expected peaks for Bi 4f7/2 and Bi 4f5/2 at 157.5 and 162.8 eV, Sb 3d3/2 and 3d5/2 at 538.3 and 528.8 eV, Te 3d3/2 and 3d5/2 at 582.9 and 572.5 eV (Sb spectra shown in figure 7).

These values are in good agreement with reported binding energies in  $\text{Sb}_2\text{Te}_3$  and  $\text{Bi}_2\text{Te}_3$ . In addition, a second set of peaks at higher binding energies is visible for every element, comparable to the binding energies in  $\text{Bi}_2\text{O}_3$ ,  $\text{Sb}_2\text{O}_3$  and  $\text{TeO}_2$ . Since only one O 1s peak at 530.3 eV was observed, the film surface is most likely covered with a thin  $(\text{Bi}_{0.5}\text{Sb}_{0.5})_2\text{Te}_{3-x}\text{O}_x$  layer, which is expected since the film was handled and transferred under ambient conditions, leading to post-oxidation (surface oxidation) of the material. This was proven by an elemental depth profile analysis, showing a strong decrease of the oxidized species after a short sputtering period. After a sputtering time of 3 min, with an acceleration voltage of 3 kV corresponding to an abrasion of about 10 nm, equivalent to 10 quintuple layers, the intensity of peaks related to oxygen or carbon are below the detection limit<sup>1</sup>. The depth profile shows a nearly constant elemental composition along the growth direction in the bulk, whereas the surface is slightly tellurium rich. This is in good agreement with our previous reported measurements by scanning Auger electron microscopy (SAES), indicating a thin Te coverage of

<sup>1</sup> Note that the abrasion of material by sputtering is calibrated on silicon; no data are available for  $(\text{Bi}_x\text{Sb}_{1-x})_2\text{Te}_3$  so the given value is a rough approximation.



**Figure 7.** X-ray photoelectron spectrum of the  $(\text{Bi}_{0.5}\text{Sb}_{0.5})_2\text{Te}_3$  film grown on  $\text{Al}_2\text{O}_3(0001)$  with a native oxidized surface (black) and after 1.5 min (5 nm, red) and 3 min sputtering times (10 nm, blue).

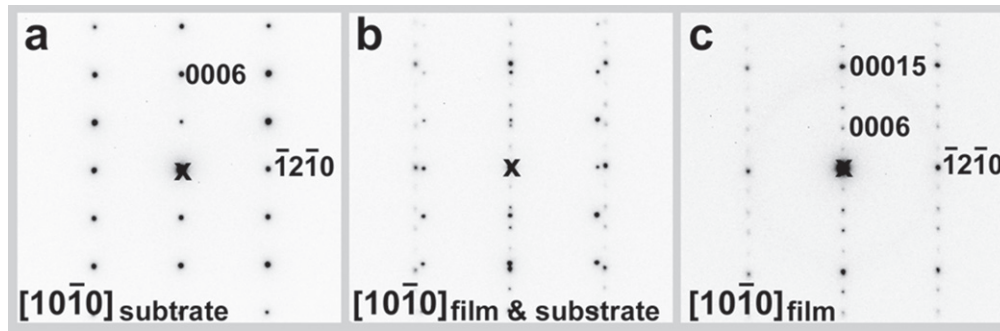


**Figure 8.** Cross-sectional TEM bright field images of a  $(\text{Bi}_{0.5}\text{Sb}_{0.5})_2\text{Te}_3$  film grown at 400 °C on  $\text{Al}_2\text{O}_3(0001)$  substrate (S: substrate, F: film, Pt: platinum layer).

the surface caused by the enhanced retention time of the less volatile tellurium precursor compared to the antimony precursor inside the reactor [52].

A TEM lamella was prepared by the FIB technique to inspect the film–substrate system in cross section. Unfortunately, the sample shifted during the milling process, so the minimal thickness of the lamella could not be reached.

However, TEM investigations show the  $(\text{Bi}_{0.5}\text{Sb}_{0.5})_2\text{Te}_3$  film with a uniform thickness of 450 nm and a flat surface on the  $\text{Al}_2\text{O}_3(0001)$  substrate (figure 8(a)). In addition, both platinum layers, which were deposited during the FIB process in order to protect the film, are visible, thereby the completeness of the film is ensured. An image at a higher magnification (figure 8(b)) reveals a sharp interface between the film and substrate showing a Fresnel contrast; an interfacial phase was not observed. Despite the unfavorable thickness of the sample, lattice fringes with a spacing of 1 nm parallel to the interface are imaged in the film. The spacing agrees with the  $d$  value of the {0003} lattice planes of  $(\text{Bi}_{0.5}\text{Sb}_{0.5})_2\text{Te}_3$ . The lattice misfit for  $\text{Al}_2\text{O}_3$  and  $(\text{Bi}_{0.5}\text{Sb}_{0.5})_2\text{Te}_3$  was calculated from structure data (literature values) to 5.6% for the d-spacings  $\{\bar{1}2\bar{1}0\}$  and 9.3% for the corresponding



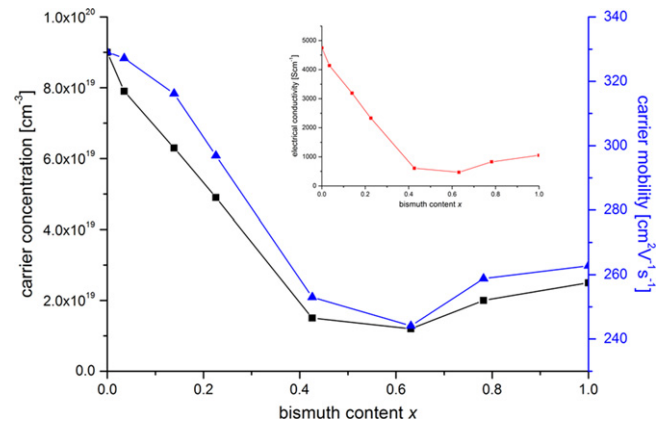
**Figure 9.** SAED pattern of the  $\text{Al}_2\text{O}_3(0001)$  substrate (a), the interface area (b), and the  $(\text{Bi}_{0.5}\text{Sb}_{0.5})_2\text{Te}_3$  film (c) grown at  $400^\circ\text{C}$  on  $\text{Al}_2\text{O}_3(0001)$ . The diffraction patterns are given with inverted contrast.

d-spacings along the  $c^*$ -axis  $\{0006\}$  in  $\text{Al}_2\text{O}_3$  and  $\{00015\}$  in  $(\text{Bi}_{0.5}\text{Sb}_{0.5})_2\text{Te}_3$ . Thus, misfit dislocations are to be expected at the interface, but were not detected in this sample; maybe such contrasts are subdued by the thickness of the sample, although the bending contours in the substrate indicate some strain generated at the interface.

The selected area electron diffraction (SAED) pattern is shown in figure 9 for the substrate (a), the interface area (b) and the film (c) along the  $[10\bar{1}0]$  axis of  $\text{Al}_2\text{O}_3$  and  $(\text{Bi}_{0.5}\text{Sb}_{0.5})_2\text{Te}_3$ , respectively. The diffraction patterns are rotated equivalent to the orientation of the interface in figure 7. The  $000l$  type reflections of the film and the substrate are in a parallel direction, thus the close packed layers of  $\text{Al}_2\text{O}_3$  and  $(\text{Bi}_{0.5}\text{Sb}_{0.5})_2\text{Te}_3$  propagate across the interface, as was found by XRD results.

The in-plane orientation shows a perfect ‘hexagon on hexagon’ orientation relationship. The  $\{\bar{1}2\bar{1}0\}$  lattice planes of the substrate and film coincide with a small mismatch in the  $\langle 10\bar{1}0 \rangle$  direction as well as the  $\{01\bar{1}0\}$  lattice planes of the film and substrate were observed to match in the  $\langle 2\bar{1}\bar{1}0 \rangle$  zone axis. The distance of the Bragg spots from the origin of the reciprocal lattice are smaller in the substrate than in the film and thus the d-spacings of the film are larger compared to the substrate, which agrees with the larger in-plane lattice parameters of the substrate.

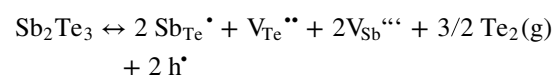
The in-plane transport properties of the MOCVD grown  $(\text{Bi}_x\text{Sb}_{1-x})_2\text{Te}_3$  films were determined by Hall effect measurements using the van der Pauw geometry in order to determine how the carrier concentration  $n$ , the carrier mobility  $\mu$ , and the conductivity  $\sigma$ , are affected by the bismuth content. For pure  $\text{Sb}_2\text{Te}_3$  films, a high carrier mobility of about  $330\text{ cm}^2\text{ V}^{-1}\text{ s}^{-1}$  was found, comparable to values measured for bulk single crystals [64] as well as other high quality  $\text{Sb}_2\text{Te}_3$  films grown by atomic layer deposition (ALD), MOCVD and MBE [45, 65, 66]. It is noteworthy, that the carrier mobility in layered antimony or bismuth telluride films is greatly enhanced compared to polycrystalline films, pointing to a preferred transport along the  $ab$ -plane, perpendicular to the  $c$ -axis. As shown in figure 10, with increasing bismuth content  $x$ , the carrier mobility decreases to a minimum value of  $244\text{ cm}^2\text{ V}^{-1}\text{ s}^{-1}$  for  $x=0.63$ . With further increasing bismuth content, the mobility increases up to  $260\text{ cm}^2\text{ V}^{-1}\text{ s}^{-1}$  for pure  $\text{Bi}_2\text{Te}_3$ . Comparable results were



**Figure 10.** Carrier concentration and mobility of  $(\text{Bi}_x\text{Sb}_{1-x})_2\text{Te}_3$  films grown at  $400^\circ\text{C}$  on  $\text{Al}_2\text{O}_3(0001)$ .

observed by L-P Hu *et al* for hot pressed bulk  $(\text{Bi}_x\text{Sb}_{1-x})_2\text{Te}_3$  [67]. A Kadhim *et al* reported similar behavior of carrier mobility for samples with high bismuth concentrations for bulk  $(\text{Bi}_x\text{Sb}_{1-x})_2\text{Te}_3$  samples obtained by the solid-state microwave route [68]. They explained this effect by the decreased concentration of lattice defects, which leads to a subdued influence of the ‘alloy scattering’ resulting in increasing carrier mobility. Zhang *et al* reported on comparably high values for the carrier mobility for highly textured  $\text{Sb}_2\text{Te}_3$  films with a maximum value of  $305\text{ cm}^2\text{ V}^{-1}\text{ s}^{-1}$  for 121 nm thick films grown by MBE on  $\text{Si}(111)$  [65]. For ALD fabricated films, Niensch *et al* found a similar value of about  $270\text{ cm}^2\text{ V}^{-1}\text{ s}^{-1}$  [66]. These findings are significantly higher compared to polycrystalline samples, for which Giani *et al* in their pioneering work reported the best value of  $196\text{ cm}^2\text{ V}^{-1}\text{ s}^{-1}$  [69].

The pure  $\text{Sb}_2\text{Te}_3$  film shows a high  $p$ -type carrier concentration of  $p=9.0 \cdot 10^{19}\text{ cm}^{-3}$ . With increasing  $x$ ,  $p$  drops to the minimum value of  $1.5 \cdot 10^{19}\text{ cm}^{-3}$  for  $x=0.63$ . For Bi-rich films,  $p$  increases slightly to the maximum value of  $3.0 \cdot 10^{19}\text{ cm}^{-3}$  for pure  $\text{Bi}_2\text{Te}_3$ . The origin of the holes is the formation of antisite defects, i.e. the occupation of Te sites by Sb or Bi atoms in the tetradymite lattice, acting as acceptors. This can be expressed by the following equation, which is also valid for  $\text{Bi}_2\text{Te}_3$ : [70]



For  $\text{Bi}_2\text{Te}_3$ , a second equation for the occupation of Bi sites by Te atoms leading to n-type material, can be formulated. The formation energy of antisite defects increases with increasing bond polarity. Based on the electronegativity  $\chi$  (Pauling) for Bi(1.9), Sb(2.05) and Te(2.10), the formation energy for antisite defects is 0.35 eV for  $\text{Sb}_2\text{Te}_3$  and 0.5 eV for  $\text{Bi}_2\text{Te}_3$ . Whereas  $\text{Sb}_2\text{Te}_3$  is always p-type,  $\text{Bi}_2\text{Te}_3$  can be p-type with excess Te and n-type with excess Bi.

## Conclusion

We successfully grew epitaxial  $(\text{Bi}_x\text{Sb}_{1-x})_2\text{Te}_3$  films on commercially available, inexpensive  $\text{Al}_2\text{O}_3(0001)$  substrates by the MOCVD process using tailor-made metal organic precursors. The as-grown  $(\text{Bi}_x\text{Sb}_{1-x})_2\text{Te}_3$  films are crystalline and show a very smooth surface morphology as was proven by XRD, SEM, AFM and TEM. The transport properties strongly depend on the film composition, and the carrier concentration can be tuned in a wide range. This is of particular interest for the thermoelectric properties of the films, since the Seebeck coefficient, as well as the electrical and thermal conductivity, strongly depend on the carrier concentration.

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